

Gas Transport and Exchange: Interaction Between O₂ and CO₂ Exchange

CJ Brauner and JL Rummer, University of British Columbia, Vancouver, BC, Canada

© 2011 Elsevier Inc. All rights reserved.

Introduction

Basis for the Interaction between O₂ and CO₂:

Bohr–Haldane Effect

Theoretical Bohr–Haldane Coefficient Optimal for Oxygen Delivery

Nonlinear Bohr–Haldane Effect within the OEC
Implications of Non-Steady-State Conditions for the Interaction between O₂ and CO₂ Exchange

Concluding Remarks

Further Reading

Glossary

Arterial Adjective pertaining to blood that has been oxygenated.

Bohr effect Effect of the proton concentration (pH) on the oxygen affinity of hemoglobin.

Carbonic anhydrase A zinc metalloenzyme that reversibly catalyzes the reaction between CO₂ and H₂O to form H⁺ and HCO₃⁻.

Equilibrium Pertaining to the situation when all forces acting are balanced by others resulting in a stable unchanging system.

Haldane effect Proton binding to hemoglobin (as a function of oxygenation).

Hypercarbia High levels of carbon dioxide in water or air.

P₅₀ The oxygen partial pressure at half-maximal oxygen saturation of blood or hemoglobin.

Partial pressure The pressure that one gas would have if it alone occupied the same volume at the same temperature as the mixture.

Root effect Property of hemoglobins in some fishes such that, in the presence of acid, it is impossible for hemoglobin molecules to be completely saturated with oxygen, even at extremely high oxygen partial pressures.

Venous Adjective pertaining to blood that has passed through tissues having had some oxygen removed for metabolism.

Introduction

All animals produce about the same amount of CO₂ as O₂ consumed through the process of metabolism. O₂ is taken up from the environment and delivered to the tissues by the blood, and CO₂ is released by the tissues and transported by the blood for release to the environment. Hemoglobin (Hb), encapsulated within the red blood cell (RBC), plays a vital role in both O₂ and CO₂ transport in the blood of all vertebrates (with the exception of icefish, the only vertebrate lacking Hb). This article focuses on the nature of the interaction between O₂ and CO₂ at the level of Hb, a topic that has been well studied since the dual role of Hb was discovered in the early 1900s.

Oxygen uptake at the gas-exchange organ (referred to as the gills from this point forward, although the skin and various air-breathing organs can contribute to gas exchange in some fishes) facilitates CO₂ removal through the Haldane effect, where Hb-oxygenation releases H⁺s that combine with HCO₃⁻ to form CO₂, which diffuses into the environment (see also **Transport and Exchange**

of Respiratory Gases in the Blood: Carbon Dioxide Transport and Excretion, Carbonic Anhydrase in Gas Transport and Exchange, and Red Blood Cell Function). CO₂ removal at the gills and the associated increase in blood pH results in an increase in Hb-O₂ affinity, which increases the driving force for O₂ diffusion across the gills, ultimately facilitating O₂ uptake via the Bohr effect (see also **Transport and Exchange of Respiratory Gases in the Blood:** Evolution of the Bohr Effect). At the tissues, the reverse occurs. CO₂ diffusion into the blood creates an acidosis that facilitates O₂ delivery to the actively metabolizing tissues via the Bohr effect. Oxygen delivery facilitates CO₂ uptake by Hb, and thus CO₂ removal from the tissues via the Haldane effect. Thus, there is an intimate interaction between O₂ and CO₂ transport at both the gills and the tissues at the level of Hb within the RBC, which is determined in part by the magnitude of the Bohr and Haldane effects.

Tremendous diversity in Hb characteristics exists within the fishes, making this group of animals particularly interesting for investigating the interaction between O₂ and CO₂ exchange. Some fish possess very small, even

nonexistent Bohr and Haldane effects with relatively high Hb-buffer values, substantially limiting the interaction between O₂ and CO₂ exchange. However, the majority of fishes (teleosts) possess large Bohr and Haldane effects (as well as a Root effect, where oxygen-carrying capacity of the blood is reduced at low pH) and low Hb-buffer values, collectively resulting in tightly coupled O₂ and CO₂ exchange. Furthermore, many teleost fishes exhibit a nonlinear Bohr and Haldane effect over the region of the oxygen-equilibrium curve (OEC), which has further implications for the nature of the interaction between O₂ and CO₂ exchange.

Basis for the Interaction between O₂ and CO₂: Bohr–Haldane Effect

The interaction between O₂ and CO₂ exchange is largely determined by the Bohr and Haldane effects as discussed above; however, their respective magnitudes are important in determining the nature of the interaction. The Bohr effect describes how the affinity of Hb for O₂ is affected for a given change in the H⁺ concentration (pH) of the blood. It is calculated as follows:

$$\text{Bohr coefficient} = -\Delta \log P_{50} / \Delta \text{pH} \quad [1]$$

where P_{50} refers to the partial pressure of O₂ (P_{O_2}) at which 50% of the Hb molecules are oxygenated.

The Haldane effect describes how the affinity of Hb for H⁺'s and CO₂ is affected by changes in Hb-O₂ saturation. It is calculated as follows:

$$\text{Haldane coefficient} = \Delta \text{H}^+ \quad [2]$$

where ΔH^+ refers to the moles of H⁺'s released per mole of O₂ bound to Hb.

Although the Bohr and Haldane effects are often discussed in terms of their respective roles relative to O₂ and CO₂ dynamics at the level of the Hb, they are actually mirror images of the same phenomenon. While the Bohr effect describes changes in Hb-O₂ affinity that arise from a change in H⁺ concentration, the Haldane effect describes the changes in Hb-H⁺ affinity that arise from a change in P_{O_2} , and therefore Hb-O₂ saturation. Thus, the Bohr and Haldane effects are linked functions, as has been recognized by the classic Wyman linkage equation:

$$(\log P_{O_2} / \text{pH}) Y = (\text{H}^+ / Y) \text{pH} \quad [3]$$

where Y refers to Hb-O₂ saturation and H⁺ refers to the number of protons bound per heme molecule. Assuming that the shape of the OEC is symmetrical and H⁺ release is linear with Hb-binding O₂, which is often the case in vertebrates (but not always, as described below in section Nonlinear Bohr–Haldane Effect), the linkage equation is often reduced to the following:

$$-\Delta \log P_{50} / \Delta \text{pH} = \Delta \text{H}^+ \quad [4]$$

The point is that the Bohr and Haldane coefficients are numerically equivalent, and will be referred to as the Bohr–Haldane coefficient, and reported as a positive value from this point forward. Further, this relationship has been experimentally validated. Air-breathing animals typically have moderate Bohr–Haldane coefficients (i.e., 0.35), while most teleosts have relatively large Bohr–Haldane coefficients (0.5 to >1.0). The numeric value has large implications for the nature of the interaction between O₂ and CO₂ exchange *in vivo*, as described in the following section.

Theoretical Bohr–Haldane Coefficient Optimal for Oxygen Delivery

The potential benefit to tissue-O₂ delivery associated with the Bohr effect is quantified as the product of the pH change associated with metabolic CO₂ production during blood transit through a tissue (arterial–venous pH change (pH_{a-v})) and the magnitude of the Bohr–Haldane coefficient. A large Bohr effect is often assumed to convey a greater benefit to tissue O₂ delivery, but this will only be true if the pH change described above is sufficient. Associated with a large Bohr effect is a large Haldane effect; therefore, upon deoxygenation, Hb will bind H⁺'s, thereby reducing the magnitude of the pH_{a-v} and consequently the expression of the Bohr effect at the tissues. In 1983, Lapennas conducted an analysis to determine the optimal Bohr coefficient for O₂ delivery under steady-state conditions. Assuming that the pH_{a-v} arises from tissue CO₂ production (and associated conversion to HCO₃⁻ and H⁺) and that most animals have a tissue respiratory quotient (RQ) between 0.7 and 1.0 (moles of CO₂ produced per mole of O₂ consumed; see also **Tissue Respiration: Cellular Respiration**), a Bohr–Haldane coefficient of 0 will result in the largest pH change within the tissues. This would be due to the absent Haldane effect and associated H⁺ binding upon deoxygenation, but would have no effect on O₂ delivery due to the lack of pH sensitivity of the Hb (i.e., no Bohr effect). Conversely, a Bohr–Haldane coefficient equivalent to the RQ (e.g., 1.0) will result in no pH change due to the increased H⁺ binding associated with the large Haldane effect. That is, all protons produced by CO₂ production at the tissues will be bound by Hb as O₂ is released to the tissues, and, despite the presence of a pH-sensitive Hb with a large Bohr effect, there may be no benefit to O₂ delivery due to the lack of pH_{a-v} . Furthermore, a Bohr–Haldane coefficient greater than RQ may result in a reverse pH_{a-v} , during blood transit, which actually could impair tissue-O₂ delivery.

Lapennas determined that the optimal Bohr coefficient for O₂ delivery under steady-state conditions (and with

many assumptions) is $0.5 \times RQ$. This represents a compromise between pH sensitivity of the Hb and the resulting pH change that occurs during capillary blood transit. Because many air-breathing vertebrates have Bohr–Haldane coefficients of 0.35 (which is very close to optimal if RQ is assumed to be 0.7), he concluded that their Hbs have been optimized for O₂ delivery (**Figure 1**). Given that most teleost fish possess Bohr–Haldane coefficients much greater than $0.5 \times RQ$ (see also **Transport and Exchange of Respiratory Gases in the Blood: Evolution of the Bohr Effect**), it has been assumed that, under steady-state conditions in most tissues, fish Hbs may be optimized for CO₂ transport and acid–base homeostasis rather than tissue-O₂ delivery. This clearly does not apply to the unique structures within the swimbladder and eye, where there exists a tremendous potential for generating and localizing an acidosis, which in conjunction with the Root effect and associated large Bohr–Haldane coefficients generates incredibly high O₂ tensions (see also **Transport and Exchange of Respiratory Gases in the Blood: Root Effect: Molecular Basis, Evolution of the Root Effect and Rete Systems and Root Effect: Root Effect**

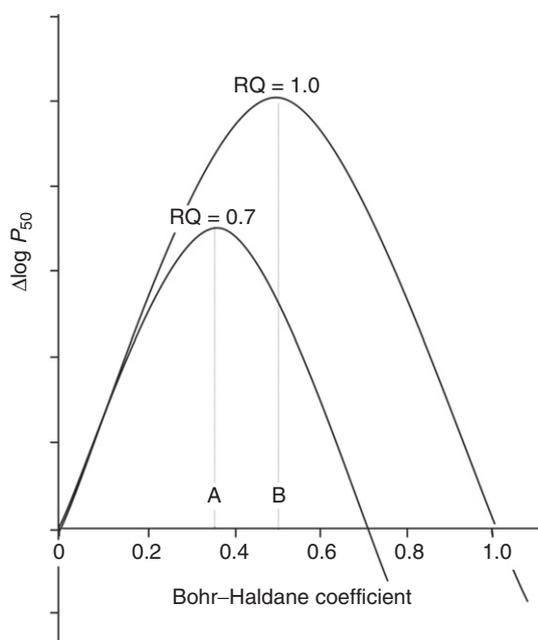


Figure 1 The optimal Bohr–Haldane coefficient: theoretical Bohr shifts, as described by a change in P_{50} ($\Delta \log P_{50}$) during blood capillary transit using two respiratory quotients (RQs). Units have been omitted intentionally from the y-axis, because the magnitude of this response will vary by species, depending on Hb buffer values. A and B indicate Bohr–Haldane coefficients optimal for O₂ delivery for RQ values of 0.7 and 1.0, respectively. Each curve crosses the x-axis at both zero and the RQ, two points at which Lapennas suggests there will be no benefit to O₂ delivery associated with the Bohr–Haldane effect. Modified from Lapennas GN (1983) The magnitude of the Bohr coefficient: Optimal for oxygen delivery. *Respiration Physiology* 54(2): 161–172.

Definition, Functional Role in Oxygen Delivery to the Eye and Swimbladder). However, in other tissues, given Lapennas' assumptions, a very large Bohr–Haldane coefficient would not benefit tissue-oxygen delivery. Despite Lapennas' many assumptions, his analyses serve as an interesting framework for hypothesizing how different Bohr–Haldane coefficients within and between species may influence the interaction between O₂ and CO₂ exchange *in vivo*.

Nonlinear Bohr–Haldane Effect within the OEC

The assumption for many models depicting vertebrate Hb function is that the magnitude of the Bohr–Haldane coefficient is relatively constant across the entire OEC. Although this may be true for most air-breathing vertebrates, it is not the case in several fish species, where most of the Bohr–Haldane effect occurs in the upper reaches of the OEC (between 50% and 100% Hb-O₂ saturation) with very little expression below 50% Hb-O₂ saturation. The nonlinear Bohr–Haldane effect is typically associated with species that possess Root-effect Hbs, perhaps suggesting that nonlinearity is common among teleosts. When fish use different regions of the OEC for gas exchange *in vivo*, implications for a nonlinear Bohr–Haldane effect influencing the interaction between O₂ and CO₂ exchange become interesting.

The entire Bohr–Haldane effect may be exploited in resting fish, where venous Hb-O₂ saturation levels rarely fall below 50%. In resting rainbow trout *Oncorhynchus mykiss*, the Bohr–Haldane coefficient calculated over the region of the OEC used *in vivo* is approximately 1.0, a value close to RQ. Therefore, CO₂ excretion at the gills and CO₂ and acid–base transport at the tissues will be maintained because Hb will bind all H⁺'s released from the tissues during O₂ delivery. However, when fish are forced to swim, the arterial–venous O₂ difference increases, requiring a greater region of the OEC to be used for gas exchange. The magnitude of the Bohr–Haldane coefficient calculated over the region of the OEC used for gas exchange during exercise is reduced to a value of 0.4–0.5, remarkably close to the value deemed optimal for O₂ delivery by Lapennas. Accordingly, during periods of increased activity and therefore muscle-O₂ demand, the nonlinear Bohr–Haldane effect may be important for optimizing O₂ delivery. Thus, the nature of the interaction between O₂ and CO₂ exchange, if a nonlinear Bohr–Haldane effect is present, changes with exercise intensity and the region of the OEC exploited for gas exchange.

Implications of Non-Steady-State Conditions for the Interaction between O₂ and CO₂ Exchange

Most of the discussion to this point has assumed steady-state, equilibrium conditions; however, it is unlikely that such conditions exist *in vivo*, because gas exchange consists of a complex combination of blood flow, boundary layers, chemical reactions, and diffusion. For example, if CO₂ from the tissues diffuses into the blood faster than O₂ diffuses to the tissues, a large Bohr–Haldane coefficient could facilitate O₂ delivery during blood capillary transit. The only way to determine whether this occurs *in vivo* is by direct measurement, which is very difficult. In the only published study to date where this has been conducted in fish, P_{O_2} was measured in real time via an optode implanted in the red muscle of resting rainbow trout. Results indicate a much higher tissue P_{O_2} than in mammals. Despite an *in vivo* Bohr–Haldane coefficient of 1.0 under resting conditions (far greater than the optimal value determined as described above), it could be that general O₂ delivery is enhanced in rainbow trout, and perhaps other fish species. Clearly, additional studies of this nature are required to investigate this further.

In fish, the greatest disequilibrium state at the level of the RBC is likely associated with catecholamine-stimulated Na⁺/H⁺ exchange. During stress, or when Hb-O₂ saturation falls below 50%, metabolic CO₂ and H⁺ production may be elevated. Catecholamines such as adrenaline and noradrenaline are released into the circulation and bind to β -adrenergic receptors on the RBC membrane. Through adenylate cyclase, which activates 3',5'-cyclic monophosphate (cAMP), the β -adrenergic Na⁺/H⁺ exchange (β NHE) on the RBC membrane is activated. The carbonic anhydrase (CA)-catalyzed hydration of CO₂ inside the RBC produces H⁺ and HCO₃⁻; the former are removed in exchange for Na⁺ via the β NHE to restore pH_i and therefore Hb-O₂ affinity, but at the cost of decreasing plasma pH. The HCO₃⁻ is removed via anion exchange for Cl⁻. The resulting osmotic gradient activates the Na⁺, K⁺ pump, and osmotically obliged water enters the RBC, causing the cell to swell (see also **Transport and Exchange of Respiratory Gases in the Blood: Carbon Dioxide Transport and Excretion and Red Blood Cell Function**). This process is thought to have evolved to protect O₂ uptake at the respiratory surface during a general acidosis in species possessing Root-effect Hbs, where an acidosis drastically decreases not only Hb's affinity but also carrying capacity for O₂. However, this is also an example of a disequilibrium state where O₂ and CO₂ transport dynamics do not follow steady-state models.

The adrenergically activated β NHE elevates pH_i but only in the absence of plasma-accessible CA. If CA were accessible to the plasma, this would short-circuit the RBC

β NHE, because H⁺s would combine with HCO₃⁻ to form CO₂ and back-diffuse into the RBC, decreasing pH_i (Figure 2). Fish lack plasma-accessible CA at the gills and thus adrenergically activated β NHE protects O₂ uptake at the respiratory surface; however, CA may be plasma accessible in some tissues, where the CA IV isoform (see also **Transport and Exchange of Respiratory Gases in the Blood: Carbonic Anhydrase in Gas Transport and Exchange**) is bound to endothelial cells and plasma oriented. When the β NHE is activated during stress, and blood passes through capillaries possessing CA, the RBC β NHE is short-circuited, resulting in a much larger pH_{a-v} than which would otherwise occur, greatly facilitating O₂ delivery to the tissues.

Short-circuiting of RBC β NHE has been validated *in vitro*. When CA is added to adrenergically stimulated, mildly acidified RBCs in a closed system, it results in a rapid acidification of RBC pH_i and a dramatic increase in P_{O_2} . Accordingly, short-circuiting of RBC β NHE *in vivo* could potentially have large effects. In rainbow trout implanted with an O₂ optode in red muscle, which allows real-time tissue P_{O_2} monitoring, inducing a mild acid-base disturbance by exposing fish to elevated environmental CO₂ levels (hypercarbia) results in a significant increase in tissue P_{O_2} . However, when plasma-accessible CA is subsequently inhibited following the addition of a non-membrane-permeable CA inhibitor to the blood, the hypercarbia-induced increase in tissue P_{O_2} is abolished, indicating that short-circuiting of RBC β NHE may be operational *in vivo* and may greatly facilitate O₂ delivery. Therefore, although rainbow trout have a large Bohr–Haldane effect, a large pH_{a-v} in the RBC may occur via exploiting disequilibrium states *in vivo*. Furthermore, this system would operate with every pass through the tissues, effectively harnessing the general acidosis to create a RBC pH_{a-v}, which will be localized to tissues that possess plasma-accessible CA.

Concluding Remarks

The intricate relationship between O₂ and CO₂ exchange has been described in past models that assume equilibrium conditions, which link the Bohr–Haldane relationship to the RQ in order to make predictions as to optimal O₂ delivery or CO₂ transport and acid–base homeostasis. However, equilibrium conditions probably never occur *in vivo*. In addition, teleosts possess a large Bohr–Haldane coefficient, which according to Lapennas' theory may be detrimental to tissue O₂ delivery. Yet, real-time measurements confirm elevated muscle P_{O_2} in rainbow trout, suggesting that a mechanism is in place to enhance O₂ delivery in fish. Teleosts' unique Root-effect Hbs convey a great potential for O₂ delivery, and possessing a nonlinear Bohr–Haldane effect allows different parts of the OEC to

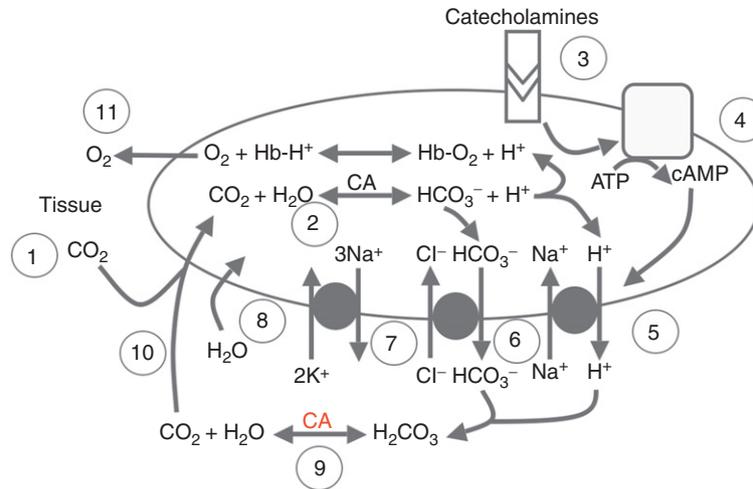


Figure 2 Short-circuiting the β -adrenergic response at the red blood cell: With the advent of an acidosis or when metabolic CO_2 production is high, the red blood cell (RBC) intracellular pH (pH_i) can decrease substantially. Metabolic CO_2 enters the RBC (1) and is catalyzed by carbonic anhydrase (CA) to form HCO_3^- and H^+ (2), the latter of which will bind hemoglobin (Hb), thus releasing O_2 . Catecholamines such as adrenaline and noradrenaline are released into the circulation to bind to receptors on the RBC membrane (β), activating adenylate cyclase and 3',5'-cyclic monophosphate (cAMP) (4), which activates the Na^+ , H^+ exchanger (βNHE) (5). As protons (H^+) are removed from the RBC in exchange for Na^+ , pH_i increases and Hb-O_2 affinity is restored. Bicarbonate (HCO_3^-) is removed from the RBC in exchange for Cl^- via the anion exchanger (6), resulting in an osmotic gradient that activates the Na^+ , K^+ pump (7). Osmotically obliged water enters the cell (8) resulting in RBC swelling. As long as βNHE rates are high and CA is not accessible to the plasma, this results in an increase in RBC pH_i . However if CA is plasma accessible (9), which may occur in the tissues, H^+ and HCO_3^- removed from the RBC would be catalyzed to form CO_2 that would back-diffuse into the RBC (10). This would short-circuit the original protective function of the βNHE mechanism and favor O_2 unloading from the Hb (11), which in the case of Root-effect Hbs would be substantial, creating enhanced O_2 delivery localized to tissues possessing plasma-accessible CA.

be utilized under different conditions. Indeed, the potential exists, and during non-steady-state conditions such as intense exercise or hypoxia exposure, adrenergic stimulation of the RBC βNHE creates the acid-base disequilibrium needed to maximize this potential. Plasma-accessible CA in select locations within metabolizing tissue may be localizing the acidosis by short-circuiting the βNHE , therefore facilitating O_2 delivery to tissues to a much greater degree than that in air-breathing vertebrates. Root-effect Hbs evolved long before the eye and swimbladder retia, structures typically associated with the unique pH-sensitive Hb, suggesting a general use for such great O_2 delivery potential (see also **Transport and Exchange of Respiratory Gases in the Blood: Root Effect: Molecular Basis, Evolution of the Root Effect and Rete Systems**). Along with the increase in and nonlinearity of the Bohr-Haldane effect, teleosts appear to greatly facilitate O_2 transport with only moderate whole-blood pH changes.

See also: **Tissue Respiration: Cellular Respiration. Transport and Exchange of Respiratory Gases in the Blood: Carbonic Anhydrase in Gas Transport and Exchange; Carbon Dioxide Transport and Excretion; Evolution of the Bohr Effect; Hemoglobin; Red Blood Cell Function; Root Effect: Molecular Basis, Evolution of the Root Effect and Rete Systems; Root Effect: Root Effect**

Definition, Functional Role in Oxygen Delivery to the Eye and Swimbladder.

Further Reading

- Berenbrink M, Koldkjaer P, Kepp O, and Cossins AR (2005) Evolution of oxygen secretion in fishes and the emergence of a complex physiological system. *Science* 307(5716): 1752–1757.
- Brauner CJ and Randall DJ (1998) The linkage between oxygen and carbon dioxide transport. In: Perry SF and Tufts B (eds.) *Fish Physiology, Volume 17: Fish Respiration*, pp. 283–319. New York: Academic Press.
- Brauner CJ, Thorarensen H, Gallagher P, Farrell AP, and Randall DJ (2000) The interaction between O_2 and CO_2 exchange in rainbow trout during graded sustained exercise. *Respiration Physiology* 119: 83–96.
- Brauner CJ, Wang T, Val AL, and Jensen FB (2001) Non-linear release of Bohr protons with haemoglobin-oxygenation in the blood of two teleost fishes: Carp (*Cyprinus carpio*) and tambaqui (*Colossoma macropomum*). *Fish Physiology and Biochemistry* 24(2): 97–104.
- Jensen FB (1989) Hydrogen ion equilibria in fish haemoglobins. *Journal of Experimental Biology* 143: 225–234.
- Lapennas GN (1983) The magnitude of the Bohr coefficient: Optimal for oxygen delivery. *Respiration Physiology* 54(2): 161–172.
- McKenzie DJ, Wong S, Randall DJ, et al. (2004) The effects of sustained exercise and hypoxia upon oxygen tensions in the red muscle of rainbow trout. *Journal of Experimental Biology* 207(21): 3629–3637.
- Nikinmaa M (1990) *Vertebrate Red Cells: Adaptations of Function to Respiratory Requirements*. Berlin: Springer.
- Tufts B and Perry SF (1998) Carbon dioxide transport and excretion. In: Perry SF and Tufts B (eds.) *Fish Physiology, Volume 17: Fish Respiration*, pp. 229–281. New York: Academic Press.
- Wyman J (1973) Linked functions and reciprocal effects in haemoglobin: A second look. *Advanced Protein Chemistry* 19: 223–286.